# MA 138 - Calculus 2 with Life Science Applications Linear Systems <br> (Section 9.1) 

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## Systems of Equations and Their Solutions

A system of equations is a set of equations that involve the same variables. A solution of a system is an assignment of values for the variables that makes each equation in the system true. To solve a system means to find all solutions of the system.

## Example 1

Show that $(0,-10)$ and $(6,8)$ are solutions of the system

$$
\left\{\begin{array}{l}
x^{2}+y^{2}=100 \\
3 x-y=10
\end{array}\right.
$$



## A Flashback

We have already encountered systems of linear equations when we were discussing partial fraction decomposition in Section 7.3.
E.g., suppose that we have to find the integral of $\frac{6 x+4}{(3 x+1)(2 x+2)}$. The method says that we need to find constants $A$ and $B$ such that

$$
\begin{aligned}
\frac{6 x+4}{(3 x+1)(2 x+2)} & =\frac{A}{3 x+1}+\frac{B}{2 x+2} \\
& =\frac{A(2 x+2)+B(3 x+1)}{(3 x+1)(2 x+2)} \\
& =\frac{(2 A+3 B) x+2 A+B}{(3 x+1)(2 x+2)}
\end{aligned}
$$

Thus, we need to solve the following equations

$$
2 A+3 B=6 \quad 2 A+B=4
$$

## Systems of Linear Equations

A system of two linear equations in two variables has the form

$$
\left\{\begin{array}{l}
a_{1} x+b_{1} y=c_{1} \\
a_{2} x+b_{2} y=c_{2}
\end{array}\right.
$$

where $a_{1}, a_{2}, b_{1}, b_{2}, c_{1}$, and $c_{2}$ are numbers. We can use two methods

- the substitution method

■ the elimination method
to solve such systems algebraically. For linear systems, though, the elimination method is usually easier.

Geometrically: The graph of a linear system in two variables is a pair of lines. Thus, from a graphic point of view, to solve the system means that we must find the intersection point(s) of the lines.

## Substitution Method

In the substitution method we start with one equation in the system and solve for one variable in terms of the other variable.

The following describes the procedure.

1. Solve for One Variable: Choose one equation and solve for one variable in terms of the other variable.
2. Substitute: Substitute the expression you found in Step 1 into the other equation to get an equation in one variable, then solve for that variable.
3. Back-Substitute: Substitute the value you found in Step 2 back into the expression found in Step 1 to solve for the remaining variable.

## Example 2 (Example 1, Section 9.1, p. 433)

Find the solution of the system of linear equations

$$
\left\{\begin{array}{l}
2 x+3 y=6 \\
2 x+y=4
\end{array}\right.
$$



## Example 3 (Example 2, Section 9.1, p. 434)

Find the solution of the system of linear equations

$$
\left\{\begin{array}{l}
4 x+2 y=6 \\
2 x+y=4
\end{array}\right.
$$



## Example 4 (Example 3, Section 9.1, p. 434)

Find the solution of the system of linear equations

$$
\left\{\begin{array}{l}
4 x+2 y=8 \\
2 x+y=4
\end{array}\right.
$$



## Number of Solutions of a Linear System in Two Variables

For a system of linear equations in two variables, exactly one of the following is true:

1. The system has exactly one solution.
2. The system has no solution.
3. The system has infinitely many solutions.


Linear system with one solution. Lines intersect at a single point.


Linear system with no solution.
Lines are parallel, so they do not intersect.


Lines coincide.

## Towards the Elimination Method

■ The substitution method of solving systems of linear equations we have employed so far works well only for systems in few variables. To solve large systems of $m$ linear equations in $n$ variables, we need to develop a more efficient method.

- The basic strategy is to transform the system of linear equations into an equivalent system of equations that has the same solutions as the original, but a much simpler form.
- In a nutshell, in the elimination method we try to combine the equations using sums or differences to eliminate one of the variables.
- We illustrate this approach in the next example. We tag all equations with labels of the form $\left(R_{i}\right) ; R_{i}$ stands for " $i$ th row. The labels will allow us to keep track of our computations.


## Example 2 (Revisited)

- We multiply the coefficients of the first equation( $\equiv$ row) by $\frac{1}{2}$

$$
\left\{\begin{array} { l } 
{ 2 x + 3 y = 6 } \\
{ 2 x + y = 4 }
\end{array} \rightsquigarrow { } ^ { \frac { 1 } { 2 } R _ { 1 } } \left\{\begin{array}{r}
x+\frac{3}{2} y=3 \\
2 x+y=4
\end{array}\right.\right.
$$

- We add -2 times the first equation to the second one

$$
\left\{\begin{array} { r l } 
{ x + \frac { 3 } { 2 } y } & { = 3 } \\
{ 2 x + y } & { = 4 }
\end{array} \rightsquigarrow { } _ { R _ { 2 } - 2 R _ { 1 } } \left\{\begin{array}{rl}
x+\frac{3}{2} y & =3 \\
-2 y & =-2
\end{array}\right.\right.
$$

- We multiply the coefficients of the second equation by $-\frac{1}{2}$

$$
\left\{\begin{array}{rl}
x+\frac{3}{2} y & =3 \\
-2 y & =-2
\end{array} \leadsto{ }_{-\frac{1}{2} R_{2}}^{x+\frac{3}{2} y}=\left\{\begin{aligned}
x & =
\end{aligned}\right.\right.
$$

- We add $-\frac{3}{2}$ the second equation to the first one

$$
\left\{\begin{array} { r l } 
{ x + \frac { 3 } { 2 } y } & { = 3 } \\
{ y } & { = 1 }
\end{array} \varliminf ^ { R _ { 1 } - \frac { 3 } { 2 } R _ { 2 } } \left\{\begin{array}{rl}
x & \frac{3}{2} \\
& y
\end{array}\right.\right.
$$

## A Shorthand Notation- The Augmented Matrix

When we transform a system of linear equations, we make changes only to the coefficients of the variables. It is therefore convenient to introduce a notation that simply keeps track of all the coefficients.

Example 2 (Revisited)

$$
\begin{aligned}
& \left\{\begin{array}{l}
2 x+3 y=6 \\
2 x+y=4
\end{array} \quad \text { an }\left[\begin{array}{ll|l}
2 & 3 & 6 \\
2 & 1 & 4
\end{array}\right]\right. \\
& {\left[\begin{array}{rr|r}
2 & 3 & 6 \\
2 & 1 & 4
\end{array}\right] \rightsquigarrow \begin{array}{ll|l}
\frac{1}{2} R_{1}
\end{array}\left[\left.\begin{array}{ll}
1 & \frac{3}{2} \\
2 & 1
\end{array} \right\rvert\, 4\right] \rightsquigarrow R_{2}-2 R_{1}\left[\begin{array}{rr|r}
1 & \frac{3}{2} & 3 \\
0 & -2 & -2
\end{array}\right]} \\
& \left.\leadsto \quad \begin{array}{ll|l}
1 & \frac{3}{2} & 3 \\
0 & 1 & 1
\end{array}\right] \quad \rightsquigarrow \quad R_{1}-\frac{3}{2} R_{2} R_{2}\left[\begin{array}{ll|l}
1 & 0 & \frac{3}{2} \\
0 & 1 & 1
\end{array}\right]
\end{aligned}
$$

## Remark about Examples 3 and 4

## Example 3

$$
\left\{\begin{array}{l}
4 x+2 y=6 \\
2 x+y=4
\end{array} \quad\left[\begin{array}{ll|l}
4 & 2 & 6 \\
2 & 1 & 4
\end{array}\right]\right.
$$

$$
\left[\begin{array}{ll|l}
4 & 2 & 6 \\
2 & 1 & 4
\end{array}\right] \rightsquigarrow{ }^{\frac{1}{4} R_{1}}\left[\begin{array}{cc|c}
1 & \frac{1}{2} & \frac{3}{2} \\
2 & 1 & 4
\end{array}\right] \rightsquigarrow{ }_{R_{2}-2 R_{1}}\left[\begin{array}{ll|l}
1 & \frac{3}{2} & 3 \\
0 & 0 & 1
\end{array}\right]
$$

## Example 4

$$
\begin{aligned}
& \left\{\begin{array}{l}
4 x+2 y=8 \\
2 x+y=4
\end{array} \quad \text {, } \quad\left[\begin{array}{ll|l}
4 & 2 & 6 \\
2 & 1 & 4
\end{array}\right]\right. \\
& {\left[\begin{array}{ll|l}
4 & 2 & 8 \\
2 & 1 & 4
\end{array}\right] \rightsquigarrow{ }^{\frac{1}{4} R_{1}}\left[\begin{array}{ll|l}
1 & \frac{1}{2} & 2 \\
2 & 1 & 4
\end{array}\right] \rightsquigarrow{ }_{R_{2}-2 R_{1}}\left[\begin{array}{ll|l}
1 & \frac{3}{2} & 3 \\
0 & 0 & 0
\end{array}\right]}
\end{aligned}
$$

## Example 5 (Online Homework \# 3)

Determine the value of $k$ for which the following system

$$
\left\{\begin{array}{rl}
-6 x-6 y & = \\
-21 x-21 y & =k \\
-15 x-15 y & =
\end{array}-5\right.
$$

is consistent ( $\equiv$ the system has solution).

## Example 6 (Online Homework \# 4)

Solve the system

$$
\left\{\begin{array}{r}
2 x+5 y=a \\
-3 x-7 y=b
\end{array}\right.
$$

## Example 7 (Online Homework \# 6)

Determine the values of $h$ and $k$ so that the following system has an infinite number of solutions

$$
\left\{\begin{array}{rlr}
x-8 y & =h \\
2 x+k y & = & -10
\end{array}\right.
$$

## Example 8

A chemist has two large containers of sulfuric acid solution, with different concentrations of acid in each container. Blending 300 mL of the first solution and 600 mL of the second solution gives a mixture that is $15 \%$ acid, whereas 100 mL of the first mixed with 500 mL of the second gives a $12 \frac{1}{2} \%$ acid mixture. What are the concentrations of sulfuric acid in the original containers?

