Basic Functions (cont'd) Transformations of Functions

FastTrack — MA 137/MA 113 — BioCalculus Functions (2): More Examples and Transformations of Functions

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Goal: We continue with more examples of basic functions. We also study how certain transformations (≡shifting, reflecting, and stretching) of a function affect its graph. This gives us a better understanding of how to graph functions.

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Basic Functions (cont'd) Transformations of Functions

Parabolas/Quadratic Functions Additional Examples Even and Odd Functions

3/30

Expressing a quadratic function in standard form helps us sketch its graph and find its maximum or minimum value. There is a **formula** for (h, k) that can be derived from the general

quadratic function as follows:

$$f(x) = ax^{2} + bx + c$$
  
$$= a\left(x^{2} + \frac{b}{a}x\right) + c$$
  
$$= a\left(x^{2} + \frac{b}{a}x + \frac{b^{2}}{4a^{2}}\right) + c - \frac{b^{2}}{4a}$$
  
$$= a\left(x + \frac{b}{2a}\right)^{2} + c - \frac{b^{2}}{4a}$$

Thus:

$$h = -\frac{b}{2a} \qquad k = \frac{4ac - b^2}{4a}$$

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Vertex (h, k)

(Minimum)

ĥ

a > 0

(Maximum)

Vertex (h, k)

a < 0

# Parabolas/Quadratic Functions

A quadratic function is a function f of the form

$$f(x) = ax^2 + bx + c$$
 ,

where a, b, and c are real numbers and  $a \neq 0$ .

The graph of any quadratic function is a parabola; it can be obtained from the graph of  $f(x) = x^2$  by elementary transformations.

Indeed, <u>by completing the square</u>, a quadratic function  $f(x) = ax^2 + bx + c$  can be expressed in the **standard form** 

$$f(x) = a(x-h)^2 + k.$$

The graph of f is a parabola with vertex (h, k); the parabola opens upward if a > 0, or downward if a < 0.

> Basic Functions (cont'd) Transformations of Functions Even and Odd Functions

### Geometric Interpretation of Completing the Square

This interpretation goes back to the Babylonian scribes, who fully used the "cut-and-paste" geometry developed by the ancient surveyors (ca. 1700 BC). Here, x, a, and b are positive as they represent lengths:



Basic Functions (cont'd) Transformations of Functions Parabolas/Quadratic Functions Additional Examples Even and Odd Functions

## The Quadratic Formula

The previous calculation actually allows us to derive the general formula for the solution of the quadratic equation:

### The Quadratic Formula

The roots  $x_1$  and  $x_2$  of the quadratic equation  $ax^2 + bx + c = 0$ , where  $a \neq 0$ , are:  $\boxed{x_{1,2} = \frac{-b \pm \sqrt{b^2 - 4ac}}{2a}}$ 

Note: The easiest method to solve a quadratic equation is by factoring it.

Use the quadratic formula only when a factorization is not readily visible.

#### Basic Functions (cont'd) Transformations of Functions Even and Odd Functions

## **Example 1 (Torricelli's Law):**

A tank holds 50 gallons of water, which drains from a leak at the bottom, causing the tank to empty in 20 minutes. The tank drains faster when it is nearly full because the pressure on the leak is greater. Torricelli's Law gives the volume of the water remaining in the tank after t minutes as

$$V(t)=50igg(1-rac{t}{20}igg)^2 \qquad 0\leq t\leq 20.$$

- (a) Find V(0) and V(20).
- (b) What do your answers to part (a) represent?
- (c) Make a table of values of V(t) for t = 0, 5, 10, 15, 20.





- Note that k > 0, because [A], [B], and R are positive.
- We assume now that the reaction occurs in a closed vessel: that is, we add specific amounts of A and B to the vessel at the beginning of the reaction and then let the reaction proceed without further additions.
- We can express the concentrations of the reactants A and B during the reaction in terms of their initial concentrations a and b and the concentration of the molecular product [AB].
- If x = [AB], then

[A] = a - x for 0 < x < a and [B] = b - x for 0 < x < b.

• The concentration of AB cannot exceed either of the concentrations of A and B.

(For example, suppose five A molecules and seven B molecules are allowed to react; then a maximum of five AB molecules can result, at which point all of the A molecules are used up and the reaction ceases. The two

B molecules left over have no A molecules to react with.) http://www.ms.uky.edu/~ma137 Lecture #2

• Consider the reaction rate of the chemical reaction  $A + B \longrightarrow AB$ 

in which the molecular reactants A and B form the molecular product AB.

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- The rate at which this reaction proceeds depends on how often A and B molecules collide.
- The law of mass action states that the rate at which this reaction proceeds is proportional to the product of the respective concentrations of the reactants. (Here, concentration means the number of molecules per fixed volume.)
- Denote the reaction rate by R and the concentration of A and B by [A] and [B], respectively. The law of mass action says that  $R \propto [A] \cdot [B]$
- Introduce the proportionality factor k. We obtain  $R = k[A] \cdot [B]$ .

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Basic Functions (cont'd) Transformations of Functions

Parabolas/Quadratic Functions Additional Examples Even and Odd Functions

• Therefore, we get

R(x) = k(a-x)(b-x) for 0 < x < a and 0 < x < b.

- The condition 0 < x < a and 0 < x < b can be written as  $0 \leq x \leq \min(a, b)$ .
- Expand the expression for R(x), to see that R(x) is indeed a polynomial function (of degree 2)

$$R(x) = k(ab - ax - bx + x^2) = kx^2 - k(a + b)x + kab$$

for 
$$0 \leq x \leq \min(a, b)$$
.

9/30

A graph of R(x), 0 < x < a, is shown for the case a < b. (We chose k = 2, a = 2, and b = 5.)

# Example 3:

Find the scaling relation between the surface area S and the volume V of a sphere of radius R.

[More precisely, show that  $S = (36\pi)^{1/3} V^{2/3}$ , that is,  $S \propto V^{2/3}$ .]

• This should also be intuitively clear: As fewer and fewer A and B molecules are in the vessel, it becomes less and less likely that they will collide to form the molecular product AB. • When  $x = a = \min(a, b)$ , the reaction rate R(a) = 0. This is the point at which all A molecules are exhausted and the reaction necessarily ceases. 11/30http://www.ms.uky.edu/~ma137 Lecture #2 http://www.ms.uky.edu/~ma137 Lecture # Recold that the volu  $S = 4\pi \left(\frac{3}{4\pi}\right)^{\frac{2}{3}}.$ of a sphere of roolins R  $V = 4_{2}\pi R^{3}$  $\left( \begin{array}{cc} 64\pi^3 & \frac{9}{16\pi^2} \end{array} \right)^{\frac{1}{3}} \cdot \sqrt{\frac{2}{3}}$ The surface area of a sphere of radius R  $S = 4\pi R^2$ . We want to write  $(36\pi)^{\nu_3} \cdot \sqrt{2^{\prime_3}}$ S as a function of V.  $5 \propto \sqrt{\frac{2}{3}}$  $V = \frac{4}{3}\pi R^3 \longrightarrow \frac{3}{4\pi} V = R^3$ FROM : i.e. so that  $R = \left(\frac{3}{4\pi}V\right)^{\frac{1}{3}}$ . Substitute in  $S = 4\pi R^2$  to get  $S = 4\pi \left[ \left( \frac{3}{4\pi} \vee \right)^{\frac{1}{3}} \right]^2$ 

Parabolas/Quadratic Functions

Additional Examples

2(2-x)(5-x) -----

• Notice that when x = 0 (i.e., when no AB molecules have yet

• As more and more AB molecules form and, consequently, the

formed), the rate at which the reaction proceeds is at a maximum.

concentrations of the reactants decline, the reaction rate decreases.

Even and Odd Functions

Basic Functions (cont'd)

Transformations of Functions



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Basic Functions (cont'd) Transformations of Functions Vertical Shifting Horizontal Shifting Reflecting Graphs Vertical Stretching and Shrinking Horizontal Shrinking and Stretching

### Example 10:



Basic Functions (cont'd) Transformations of Functions Vertical Shifting Horizontal Shifting Reflecting Graphs Vertical Stretching and Shrinking Horizontal Shrinking and Stretching

# Example 10 (cont'd):

Sketch the graph of y = f(-x).

Sketch the graph of y = -f(x).

Sketch the graph of y = -f(-x).



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